# ELEMENTS by ICP (Aqua Regia Ashing)

MW: Table 1 CAS: Table 2 RTECS: Table 2

METHOD: 7301, Issue 1 EVALUATION: PARTIAL Issue 1: 15 March 2003

OSHA: Table 2 PROPERTIES: Table 1

NIOSH: Table 2 ACGIH: Table 2

**SAMPLE** 

BIAS:

ELEMENTS: aluminum\* calcium lead\* phosphorus thallium zinc antimony\* chromium\* lithium potassium tin zirconium\*

titanium arsenic cobalt magnesium selenium barium copper manganese silver tungsten\* molybdenum strontium vanadium beryllium iron\* cadmium lanthanum nickel tellurium yttrium

\* Some compounds of those elements require special sample treatment.

SAMPLING MEASUREMENT

SAMPLER: FILTER TECHNIQUE: INDUCTIVELY COUPLED ARGON

(0.8-µm, cellulose ester membrane, or PLASMA, ATOMIC EMISSION 5.0-µm, polyvinyl chloride membrane) SPECTROSCOPY (ICP-AES)

FLOW RATE: 1 to 4 L/min ANALYTE: Elements above

VOL-MIN: Table 1
-MAX: Table 1

ASHING
REAGENTS: Aqua regia (1 HNO<sub>3</sub>: 3 HCl)

SHIPMENT: Routine CONDITIONS: Room temperature, 30 min; 150 °C to

near dryness

STABILITY: Stable FINAL

Not determined

SOLUTION: 5% aqua regia, 25 mL
BLANKS: 2 to 10 field blanks per set

**WAVELENGTH:** Depends upon element, Table 3

ACCURACY BACKGROUND CORRECTION:

RANGE STUDIED: Not determined CORRECTION: Spectral wavelength shift

CALIBRATION: Elements in 5% aqua regia

OVERALL PRECISION (Ŝ, t): Not determined RANGE: Varies with element [1]

ACCURACY: Not determined

PRECISION (\$<sub>r</sub>): Tables 3 and 4

**APPLICABILITY:** The working range of this method is 0.005 to 2.0 mg/m³ for each element in a 500-L air sample. This is simultaneous elemental analysis, not compound specific. Verify that the types of compounds in the samples are soluble with the ashing procedure selected. This method does not digest PVC filters completely.

**ESTIMATED LOD:** 

Tables 3 and 4

**INTERFERENCES:** Spectral interferences are the primary interferences encountered in ICP-AES analysis. These are minimized by judicious wavelength selection, interelement correction factors and background correction [1-4].

**OTHER METHODS:** Flame atomic absorption spectroscopy (e.g., Methods 70XX) is an alternate analytical technique for many of these elements. Graphite furnace AAS (e.g., 7102 for Be, 7105 for Pb) is more sensitive. NIOSH Methods 7300 & 7302 are alternative digestion procedures.

## **REAGENTS:**

- 1. Nitric acid (HNO<sub>3</sub>), conc.\*, ultra pure.
- 2. Hydrochloric acid (HCI), conc.\*, ultra pure.
- Ashing acid (Aqua Regia): 1:3 (v/v) HNO<sub>3</sub>:HCI. Mix 1 volume conc. HNO<sub>3</sub> with 3 volumes conc. HCI
- Calibration stock solutions, 1000 μg/mL.
   Commercially available, or prepared per instrument manufacturer's recommendation (see step 12).
- 5. Dilution acid, 1% HNO<sub>3</sub>, 3% HCl. Add 50 mL ashing acid to 600 mL water; dilute to 1 L.
- 6. Argon.
- 7. Distilled, deionized water.
  - See SPECIAL PRECAUTIONS.

#### **EQUIPMENT:**

- Sampler: cellulose ester membrane filter, 0.8-µm pore size; or polyvinyl chloride (PVC) membrane, 5.0-µm pore size; 37-mm diameter, in cassette filter holder.
- 2. Personal sampling pump, 1 to 4 L/min, with flexible connecting tubing.
- Inductively coupled plasma-atomic emission spectrometer, equipped as specified by the manufacturer for analysis of elements of interest.
- 4. Regulator, two-stage, for argon.
- 5. Beakers, Phillips, 125-mL, or Griffin, 50-mL, with watchglass covers.\*\*
- 6. Volumetric flasks, 10-, 25-,100-mL., and 1-L\*\*
- 7. Assorted volumetric pipets as needed.\*\*
- 8. Hotplate, surface temperature 150 °C.
  - \*\* Clean all glassware with conc. nitric acid and rinse thoroughly in distilled water before use.

**SPECIAL PRECAUTIONS:** Concentrated acids are powerful oxidizers, toxic, and corrosive liquids. Wear protective clothing and work in a fume hood.

#### **SAMPLING:**

- 1. Calibrate each personal sampling pump with a representative sampler in line.
- 2. Sample at an accurately known flow rate between 1 and 4 L/min for a total sample size of 200 to 2000 L (see Table 1) for TWA measurements. Do not exceed a filter loading of approximately 2 mg total dust.

### **SAMPLE PREPARATION:**

- 3. Open the cassette filter holders and transfer the samples and blanks to clean beakers.
- 4. Add 5 mL ashing acid. Cover with a watchglass. Let stand 30 min at room temperature. NOTE: Start a reagent blank at this step.
- 5. Heat on hotplate (120 °C) until ca. 0.5 mL remains.
  - NOTE 1: Recovery of lead from some paint matrices may require other digestion techniques. See Method 7082 (Lead by Flame AAS) for an alternative hotplate digestion procedure or Method 7302 for a microwave digestion procedure.
  - NOTE 2: Some species of Al, Be, Co, Cr, Li, Mo, Sb, W, and Zr will not be completely solubilized by this procedure. Alternative solubilization techniques for most of these elements can be found elsewhere [5-10].
- Add 2 mL ashing acid and repeat step 5. Repeat this step until the solution is clear.
   NOTE: PVC filters will not completely dissolve after repeated additions of ashing acid.
- 7. Remove watchglass and rinse into the beaker with distilled water.
- 8. Increase the temperature to 150 °C and take the sample to near dryness (ca. 0.5 mL).
- 9. Dissolve the residue in 2 to 3 mL dilution acid.
- 10. Transfer the solutions quantitatively to 25-mL volumetric flasks.
- 11. Dilute to volume with dilution acid.

#### **CALIBRATION AND QUALITY CONTROL:**

12. Calibrate the spectrometer according to the manufacturer's recommendations.

NOTE: Typically, an acid blank and 1.0 µg/mL multielement working standards are used. The following multielement combinations are chemically compatible in 5% Aqua Regia:

- a. Al, As, Ba, Be, Ca, Cd, Co, Cr, Cu, Fe, La, In, Na
- b. Ag, K, Li, Mg, Mn, Ni, P, Pb, Se, Sr, Tl, V, Y, Zn, Sc
- c. Mo, Sb, Sn, Te, Ti, W, Zr
- d. Acid blank
- 13. Analyze a standard for every ten samples.
- 14. Check recoveries with at least two spiked media blanks per ten samples.

## **MEASUREMENT:**

- 15. Set spectrometer to conditions specified by manufacturer.
- 16. Analyze standards, samples, and blanks.

NOTE: If the values for the samples are above the range of the standards, dilute the solutions with dilution acid, reanalyze and apply the appropriate dilution factor in the calculations. If more sensitivity is required, the final sample volume may be held to 10.0 mL.

#### **CALCULATIONS:**

- 17. Obtain the solution concentrations for the sample,  $C_s$  ( $\mu g/mL$ ), and the average media blank,  $C_b$  ( $\mu g/mL$ ), from the instrument.
- 18. Using the solution volumes of sample, V<sub>s</sub> (mL), and media blank, V<sub>b</sub> (mL), calculate the concentration, C (mg/m³), of each element in the air volume sampled, V (L):

$$C = \frac{CsVs - CbVb}{V}, mg / m^3$$

NOTE:  $\mu g/L = mg/m^3$ 

## **EVALUATION OF METHOD:**

The precision and recovery data were determined at approximately 3x and 10x the instrumental detection limits on commercially prepared spiked filters [12] using 25.0 mL as the final sample volume. The precision and recovery data, instrumental detection limits, and analytical wavelengths are listed in Tables 3 and 4. In general, better recoveries were obtained from MCE filters than from PVC filters. The values in Tables 3 and 4 were determined with a Spectro Analytical Instruments model EOP operated according to manufacturer's instructions.

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#### **METHOD WRITTEN BY:**

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TABLE 1. PROPERTIES AND SAMPLING VOLUMES

	Proper	ties				
Element	Atomic	Atomic		e, L @ OSHA PEL_		
(Symbol)	Weight	MP, °C	MIN	MAX		
Silver (Ag)	107.87	961	250	2000		
Aluminum (Al)	26.98	660	5	100		
Arsenic (As)	74.92	817	5	2000		
Barium (Ba)	137.34	710	50	2000		
Beryllium (Be)	9.01	1278	1250	2000		
Calcium (Ca)	40.08	842	5	200		
Cadmium (Cd)	112.40	321	13	2000		
Cobalt (Co)	58.93	1495	25	2000		
Chromium (Cr)	52.00	1890	5	1000		
Copper (Cu)	63.54	1083	5	1000		
Iron (Fe)	55.85	1535	5	100		
Potassium (K)	39.10	63.65	5	1000		
Lanthanum (La)	138.91	920	5	1000		
Lithium (Li)	6.94	179	100	2000		
Magnesium (Mg)	24.31	651	5	67		
Manganese (Mn)	54.94	1244	5	200		
Molybdenum (Mo)	95.94	651	5	67		
Nickel (Ni)	58.71	1453	5	1000		
Phosphorus (P)	30.97	44	25	2000		
Lead (Pb)	207.19	328	50	2000		
Antimony (Sb)	121.75	630.5	50	2000		
Selenium (Se)	78.96	217	13	2000		
Tin (Sn)	118.69	231.9	5	1000		
Strontium (Sr)	87.62	769	10	1000		
Tellurium (Te)	127.60	450	25	2000		
Titanium (Ti)	47.90	1675	5	100		
Thallium (Tl)	204.37	304	25	2000		
Vanadium (V)	50.94	1890	5	2000		
Tungsten (W)	183.85	3410	50	1000		
Yttrium (Y)	88.91	1495	5	1000		
Zinc (Zn)	65.37	419	5	200		
Zirconium (Zr)	91.22	1852	5	200		

TABLE 2. EXPOSURE LIMITS, CAS #, RTECS

Element (Symbol)	CAS#	RTECS	Exposure Limits, mg/m³ (Ca = carcinogen) OSHA NIOSH ACGIH				
Silver (Ag)	7440-22-4	VW3500000	0.01 (dust, fume, metal)	0.01 (metal, soluble)	0.1 (metal) 0.01 (soluble)		
Aluminum (AI)	7429-90-5	BD0330000	15 (total dust) 5 (respirable) 5 (respirable fume) 2 (salts, alkyls)		10 (dust) 5 (powders, fume) 2 (salts, alkyls)		
Arsenic (As)	7440-38-2	CG0525000	varies	C 0.002, Ca	0.01, Ca		
Barium (Ba)	7440-39-3	CQ8370000	0.5	0.5	0.5		
Beryllium (Be)	7440-41-7	DS1750000	0.002, C 0.005	0.0005, Ca	0.002, Ca		
Calcium (Ca)	7440-70-2		varies	varies	varies		
Cadmium (Cd)	7440-43-9	EU9800000	0.005	lowest feasible, Ca	0.01 (total), Ca 0.002 (respir.), Ca		
Cobalt (Co)	7440-48-4	GF8750000	0.1	0.05 (dust, fume)	0.02 (dust, fume)		
Chromium (Cr)	7440-47-3	GB4200000	0.5	0.5	0.5		
Copper (Cu)	7440-50-8	GL5325000	1 (dust, mists) 0.1 (fume)	1 (dust) 0.1 (fume)	1 (dust, mists) 0.2 (fume)		
Iron (Fe)	7439-89-6	NO4565500	10 (dust, fume)	5 (dust, fume)	5 (fume)		
Potassium (K)	7440-09-7	TS6460000					
Lanthanum	7439-91-0		_	_			
Lithium (Li)	7439-93-2						
Magnesium (Mg)	7439-95-4	OM2100000	15 (dust) as oxide 5 (respirable)				
Manganese (Mn)	7439-96-5	009275000	C 5 1; STEL 3		5 (dust) 1; STEL 3 (fume)		
Molybdenum (Mo)	7439-98-7	QA4680000	5 (soluble) 15 (total insoluble)	5 (soluble) 10 (insoluble)	5 (soluble) 10 (insoluble)		
Nickel (Ni)	7440-02-0	QR5950000	1 0.015, Ca		0.1 (soluble) 1 (insoluble, metal)		
Phosphorus (P)	7723-14-0	TH3500000	0.1	0.1	0.1		
Lead (Pb)	7439-92-1	OF7525000	0.05	0.05	0.05		
Antimony (Sb)	7440-36-0	CC4025000	0.5	0.5	0.5		
Selenium (Se)	7782-49-2	VS7700000	0.2	0.2	0.2		
Tin (Sn)	7440-31-5	XP7320000	2	2	2		
Strontium (Sr)	7440-24-6	-	-	-			
Tellurium (Te)	13494-80-9	WY2625000	0.1	0.1	0.1		
Titanium (Ti)	7440-32-6	XR1700000					
Thallium (TI)	7440-28-0	XG3425000	0.1 (skin) (soluble)	0.1 (skin) (soluble)	0.1 (skin)		
Vanadium (V)	7440-62-2	YW240000		C 0.05			
Tungsten	7440-33-7	-	5	5 10 (STEL)	5 10 (STEL)		
Yttrium (Y)	7440-65-5	ZG2980000	1	N/A	1		
Zinc (Zn)	7440-66-6	ZG8600000	_				
Zirconium (Zr)	7440-67-7	ZH7070000	5	5, STEL 10	5, STEL 10		

TABLE 3. MEASUREMENT PROCEDURES AND DATA [1]. Mixed Cellulose Ester Filters (0.45 $\mu$ m)

Element (a)	wavelength (nm)	Est.LOD (µg/ Filter)	LOD (ng/mL)	Certified 3x LOD (µg/filter) (b)	% Recovery (c)	Percent RSD (N=25)	Certified 10x LOD (µg/filter) (b)	% Recovery (c)	Percent RSD (N=25)
Ag	328	0.042	1.7	0.77	100.3	2.39	3.21	93.4	4.95
ΑI	167	0.115	4.6	1.54	208.1	42.4	6.40	99.6	9.43
As	189	0.140	5.6	3.08	97.6	4.71	12.90	95.1	1.14
Ва	455	0.005	0.2	0.31	104.3	1.65	1.29	100.8	1.54
Ве	313	0.005	0.2	0.31	99.6	1.42	1.29	100.6	0.68
Ca	317	0.908	36.3	15.4	101.6	5.01	64.00	101.6	1.42
Cd	226	0.0075	0.3	0.31	106.8	2.60	1.29	99.2	0.76
Co	228	0.012	0.5	0.31	105.6	1.64	1.29	100.4	0.87
Cr	267	0.020	0.8	0.31	97.0	27.0	1.29	88.0	5.38
Cu	324	0.068	2.7	1.54	118.9	65.2	6.40	102.0	0.68
Fe	259	0.095	3.8	1.54	114.9	43.0	6.40	82.7	7.81
K	766	1.73	69.3	23.00	94.7	2.60	96.40	95.8	0.98
La	408	0.048	1.9	0.77	105.7	1.80	3.21	101.3	0.84
Li	670	0.010	0.4	0.31	104.3	2.37	1.29	99.3	0.89
Mg	279	0.098	3.9	1.54	105.2	4.23	6.40	99.2	1.24
Mn	257	0.005	0.2	0.31	103.5	1.64	1.29	91.2	2.01
Мо	202	0.020	8.0	0.31	108.9	2.70	1.29	97.4	1.25
Ni	231	0.020	8.0	0.31	112.2	2.28	1.29	94.2	1.73
Р	178	0.092	3.7	1.54	93.2	10.9	6.40	97.1	5.93
Pb	168	0.062	2.5	1.54	88.0	6.52	6.40	102.2	1.06
Sb	206	0.192	7.7	3.08	50.1	54.7	12.90	80.0	19.46
Se	196	0.135	5.4	2.30	93.2	8.38	9.64	89.1	7.23
Sn	189	0.040	1.6	0.77	25.8	81.9	3.21	91.7	16.39
Sr	407	0.005	0.2	0.31	100.8	1.27	1.29	99.3	0.66
Te	214	0.078	3.1	1.54	103.1	1.88	6.40	95.0	1.31
Ti	334	0.050	2.0	0.77	98.3	1.88	3.21	96.0	1.06
ΤI	190	0.092	3.7	1.54	101.3	3.57	6.40	98.2	0.71
V	292	0.028	1.1	0.77	106.0	1.38	3.21	101.3	0.81
W	207	0.075	3.0	1.54	64.9	21.8	6.40	74.1	11.34
Υ	371	0.012	0.5	0.31	104.3	1.55	1.29	99.3	0.72
Zn	213	0.310	12.4	4.60	99.8	9.73	19.30	98.0	0.86
Zr	339	0.022	0.9	0.31	52.5	71.2	1.29	76.6	18.19

<sup>(</sup>a) Bold values are qualitative only, because of poor recovery.

<sup>(</sup>b) Values are certified by Inorganic Ventures INC. at 3x and 10x the approximate instrumental LOD.

<sup>(</sup>c) Values reported were obtained with a Spectro Analytical Instruments EOP ICP; performance may vary with instrument and should be independently verified.

TABLE 4. MEASUREMENT PROCEDURES AND DATA [1]. Polyvinyl Chloride Filter (5.0  $\mu$ m)

Element (a)	wavelength (nm)	Est. LOD (µg/ Filter)	LOD (ng/mL)	Certified 3x LOD (µg/filter) (b)	% Recovery (c)	Percent RSD (N=25)	Certified 10x LOD (µg/ filter) (b)	% Recovery (c)	Percent RSD (N=25)
	000	0.040	4 7	0.70		0.0	0.40	^	04.7
Ag Al	328 167	0.042 0.115	1.7 4.6	0.78 1.56	57.9 -1.9	0.2	3.18 6.40	<b>55.0</b> 112.1	21.7 59.6
Ai As	189	0.113	4.6 5.6	3.10	-1.9 78.2	1.6	12.70	80.2	7.9
Ba	455	0.140	0.2	0.31	76.2 <b>73.0</b>	0.1	1.27	95.7	3.7
<b>ва</b> Ве	313	0.005	0.2	0.31	73. <b>0</b> 81.1	0.1	1.27	95.7 97.2	4.3
Са	313	0.003	36.3	15.60	68.2	4.9	64.00	97.2 97.7	4.5 4.5
C <b>a</b> Cd	226	0.908	0.3	0.31	86.7	0.1	1.27	97.7 97.4	4.3
Co	228	0.0073	0.5	0.31	83.8	0.1	1.27	99.2	4.4
Cr	267	0.012	0.5	0.31	80.1	0.1	1.27	99.2	6.8
Cu	324	0.020	2.7	1.56	75.9	0.1	6.40	9 <del>4</del> .1 96.1	4.3
Fe	259	0.008	3.8	1.56	73. <del>9</del> 78.4	0.5	6.40	88.4	9.0
<b>K</b>	766	1.73	69.3	23.40	61.4	3.1	95.00	91.6	5.7
	408	0.048	1.9	1.78	34.4	0.4	3.18	95.3	3.8
<b>La</b> Li	408 670	0.046	0.4	0.31	<b>34.4</b> 76.3	0.4	1.27	95.3 96.0	3.6 4.7
Mg	279	0.010	3.9	1.56	70.3 77.5	0.6	6.40	94.0	4.7
Mn	27 <i>9</i> 257	0.098	0.2	0.31	77.4	0.0	1.27	93.4	4.0
Mo	202	0.003	0.2	0.31	77. <del>4</del> 79.7	0.1	1.27	93.4 89.2	4.2 9.8
Ni	231	0.020	0.8	0.31	86.2	0.2	1.27	100.8	4.8
P	178	0.020	3.7	1.56	76.9	0.1	6.40	<b>69.0</b>	4.6 14.5
Pb	168	0.092	3.7 2.5	1.56	82.0	0.9	6.40	99.4	4.4
	206	0.002	7.7	3.10	<b>40.3</b>	1.5	12.70		4.4 76.5
<b>Sb</b> Se	196	0.192	7.7 5.4	2.30	<b>40.3</b> 89.4	1.5 1.2	9.50	<b>23.0</b> 87.5	9.9
	189	0.133	1.6	0.78	101.1	0.4	3.18	21.1	124.0
Sn Sr	407	0.040	0.2	0.78	73.4	0.4	1.27	95.2	3.9
Te	214	0.003	3.1	1.56	91.8	0.1	6.40	95.2 85.3	3.9 7.5
Ti	334	0.070	2.0	0.78	<b>53.4</b>	0.7	3.18	<b>46.3</b>	39.9
TI	190	0.030	3.7	1.56	71.6	0.2	6.40	<b>46.3</b> 86.1	9.3
V	292	0.032	1.1	0.78	77.8	0.3	3.18	96.1	9.5 4.6
w	292	0.028	3.0	1.56	51.3	0.8	6.40	29.8	47.0
Y	371	0.075	0.5	0.31	<b>51.3</b> 79.6	0.8	1.27	<b>29.8</b> 95.8	47.0 4.4
r Zn	213	0.012	12.4	4.70	79.6 80.9	2.2	19.10	95.6 94.7	4.4 4.2
∠n <b>Zr</b>	339	0.310	0.9	4.70 0.31		2.2 0.1	19.10		4.2 112.7
۷r	১১৬	0.022	0.9	0.31	46.2	U. I	1.21	39.2	112.1

<sup>(</sup>a) Bold values are qualitative only because of poor recovery.

<sup>(</sup>b) Values are certified by Inorganic Ventures INC. at 3x and 10x the approximate instrumental LOD.

<sup>(</sup>c) Values reported were obtained with a Spectro Analytical Instruments EOP ICP; performance may vary with instrument and should be independently verified.